

Petrology and geochemistry of a peridotite body in Central-Carpathian Paleogene sediments (Sedlice, eastern Slovakia)

MATÚŠ KOPPA¹, FRIEDRICH KOLLER² and MARIÁN PUTIŠ¹✉

¹Department of Mineralogy and Petrology, Faculty of Natural Sciences, Comenius University, Mlynská dolina G, 842 15 Bratislava, Slovak Republic; ✉putis@fns.uniba.sk; koppa@fns.uniba.sk

²Department of Lithospheric Research, University of Vienna, Althanstrasse 14, Vienna, Austria; friedrich.koller@univie.ac.at

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Abstract: We studied representative samples from a peridotite body situated NE of Sedlice village within the Central-Carpathian Paleogene sediments in the Central Western Carpathians. The relationship of the peridotite to the surrounding Paleogene sediments is not clear. The fractures of the brecciated peridotite margin are healed with secondary magnesite and calcite. On the basis of the presented bulk-rock and electron microprobe data, the wt. % amounts of mineral phases were calculated. Most of calculated “modal” compositions of this peridotite corresponds to harzburgites composed of olivine (~70–80 wt. %), orthopyroxene (~17–24 wt. %), clinopyroxene (<5 wt. %) and minor spinel (<1 wt. %). Harzburgites could originate from lherzolitic protoliths due to a higher degree of partial melting. Rare lherzolites contain porphyroclastic 1–2 mm across orthopyroxene (up to 25 wt. %), clinopyroxene (~5–8 wt. %) and minor spinel (<0.75 wt. %). On the other hand, rare, olivine-rich dunites with scarce orthopyroxene porphyroclasts are associated with harzburgites. Metamorphic mineral assemblage of low-Al clinopyroxene (3), tremolite, chrysotile, andradite, Cr-spinel to chromite and magnetite, and an increase of fayalite component in part of olivine, indicate low-temperature metamorphic overprint. The Primitive Mantle normalized whole-rock REE patterns suggest a depleted mantle rock-suite. An increase in LREE and a positive Eu anomaly may be consequence of interactive metamorphic fluids during serpentinization. Similar rocks have been reported from the Meliatic Bôrka Nappe overlying the Central Western Carpathians orogenic wedge since the Late Cretaceous, and they could be a potential source of these peridotite blocks in the Paleogene sediments.

Key words: peridotite, petrology, geochemistry, Slovakia.

Introduction and brief geological background

The presence of meta-ultramafic rocks within (meta)siliciclastic sediments may open new insights on the mobility of serpentinized mantle fragments within accretionary and collision wedges and their interaction with continental crust (e.g. Brandon 2004; Scambelluri et al. 2004).

Representative samples were collected for study from a 500×300 m surface occurrence of an ultramafic body to the NE from Sedlice, which forms a smaller ridge in the Šariš Highlands marked on the map as the “Dunitová Skalka” (“dunite stone” of Cambel 1951; Hovorka et al. 1985). It is situated N of the zone of pre-Carboniferous metamorphosed complexes intruded by Variscan granitoids in the Branisko and Čierna hora Mountains, in the eastern part of the Central-Carpathian Paleogene flysch belt (Marschalko 1966; Plašienka et al. 1997), in the middle of basal sandstone and conglomerate beds. This zone belongs to the Central Western Carpathians (Plašienka et al. 1997). The relationship of the serpentinized peridotite body to the surrounding Paleogene sediments is not clear. The body may be a member of a buried Mesozoic complex covered by transgressive Paleogene sediments, or an olistolith within these sediments (Hovorka et al. 1985). The geological position is shown in Fig. 1.

The peridotite body near Sedlice was the subject of interests and studies of different aspects in the past. Cambel (1951) reported enstatite dunites based on microscopic and

optical identification of rock-forming minerals. The rocks with increased Cpx content associated with Spl were classified as spinel lherzolites (Cambel 1951; Fejdi & Kolník 1988; Stankovič et al. 2007). These rocks contain a certain amount of opaque minerals, usually in paragenesis with spinel. Awaruite, millerite or pentlandite have been reported in the ultramafic rocks from Sedlice by Kantor (1955) and Rojkovič (1985). The spinel-group and ore minerals were studied by Kantor (1955), Rojkovič (1985), Rojkovič et al. (1978, 1979, 1982), Spišiak et al. (2000) and Mikuš & Spišiak (2007); serpentine-group minerals by Hovorka et al. (1980, 1985); geochemistry by Hovorka (1977); geothermometry and geobarometry by Fejdi & Kolník (1988), Radvanec (2000); exsolved pyroxenes by Stankovič et al. (2007); mesoscopic structures by Jaroš et al. (1981), and tracing the spatial distribution of the body by geophysical methods was performed by Gnojek & Kubeš (1991). Despite the previous results, evolution of the body placed within the Central Carpathian Paleogene sediments remains unclear or “exotic”.

This paper reports the results of mineralogical-petrological, and geochemical study of the Sedlice ultramafic body in the Western Carpathians in the territory of eastern Slovakia based on the study of the mineral (EMPA) and whole-rock (XRF and ICP MS) chemical compositions of representative samples. The mantle and crustal evolution, and tectonic origin of these rocks is discussed.

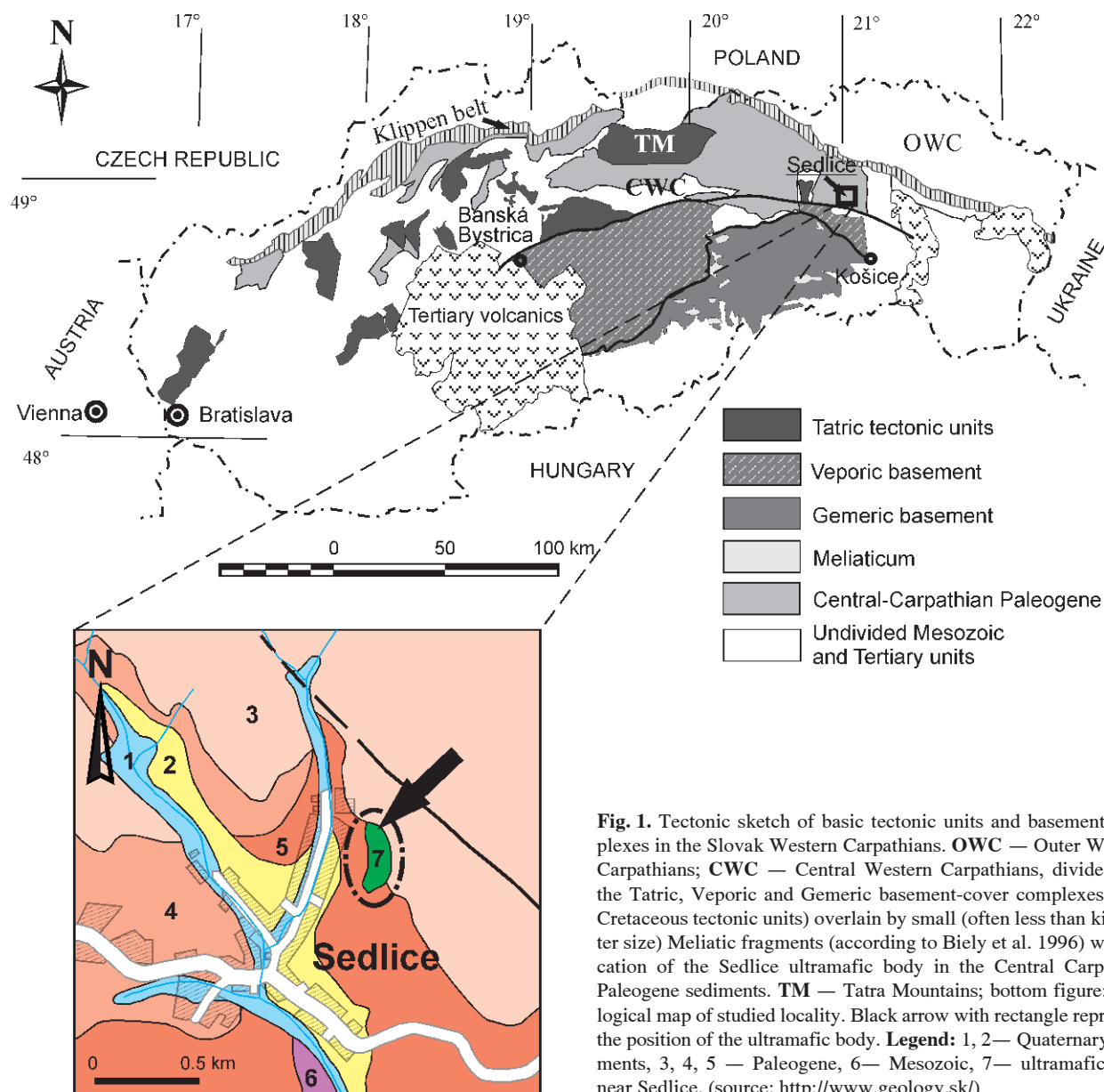


Fig. 1. Tectonic sketch of basic tectonic units and basement complexes in the Slovak Western Carpathians. **OWC** — Outer Western Carpathians; **CWC** — Central Western Carpathians, divided into the Tatric, Veporic and Gemicic basement-cover complexes (Late Cretaceous tectonic units) overlain by small (often less than kilometer size) Meliatic fragments (according to Biely et al. 1996) with location of the Sedlice ultramafic body in the Central Carpathian Paleogene sediments. **TM** — Tatra Mountains; bottom figure: Geological map of studied locality. Black arrow with rectangle represents the position of the ultramafic body. **Legend:** 1, 2— Quaternary sediments, 3, 4, 5 — Paleogene, 6— Mesozoic, 7— ultramafic body near Sedlice. (source: <http://www.geology.sk/>)

Materials and methods

X-ray fluorescence spectrometry was used for whole-rock major oxide and also some trace element concentrations contained in rock samples. In order to obtain representative major and trace element analyses, the rock samples were crushed in jaw mill and ground to the finest powder possible in an agate swing mill. Subsequently, the fused beads were prepared using $\text{Li}_2\text{B}_4\text{O}_7$ after the sample powders were dried at 110°C and heated at $950\text{--}1050^\circ\text{C}$ to determine loss on drying and ignition. Major and trace element concentrations were determined on fused beads and pressed powder pellets with a wavelength dispersive X-ray fluorescence spectrometer (PHILIPS PW 2400) at the Department of Lithospheric Research, University of Vienna. The fused beads were cast using a Philips Perl X3 automatic bead machine. A rhodium anticathode was employed for the XRF analyses.

Trace element contents (including rare earth elements — REEs) were analysed by ICP-MS at the Department of General and Analytical Chemistry, Montan-University Leoben. In total 0.1 g of fine grained sample was sintered with sodium peroxide (purity 95%) to achieve complete digestion of all silicate and spinel mineral phases (Meisel et al. 2002). Measurements were performed with an Agilent 7500 ce ICP-MS with and without He collision cell mode. The International Association of Geoanalysts (IAG) Candidate reference material MUH-1, a highly depleted serpentinized harzburgite from Kraubath, Styria, Austria (Burnham et al. 2010), was used for quality control purposes. The preceding sodium peroxide sintering and acid digesting (37% HCl) followed the procedures of Meisel et al. (2002). A solution with very accurately known concentrations of Ge, In and Re ($1\ \mu\text{l/ml}$) was employed as a reference material. The machine used was a standard quadrupole ICP-MS Agilent Technologies HP4500 with a v-groove

type Babington or a Burgener pumped nebulizer and a cooled quartz glass spray chamber. The whole-rock chemical compositions are reviewed in Table 1. The rock mineral compositions were calculated from the whole-rock and mineral chemical compositions (e.g. Bodinier & Godard 2007).

Table 1: The whole-rock major and trace element compositions of dunite (dun), lherzolites (lherz) and harzburgites (harz) from Sedlice with mineral percentual assemblages (in wt. %). **LOI** — loss on ignition, **b.d.l.** — below detection limit, **n.m.** — not measured; Cr, Ni, Zr and Pb were measured by XRF, other trace elements by ICP-MS. * — Zr value was measured by ICP-MS.

| Sample | SE-2 | SE-14 | SE-15 | SE-6b | SE-12 | SE-17b |
|------------------------------------|-------|--------|--------|--------|--------|--------|
| Rock type | dun | harz | harz | harz | lherz | lherz |
| OI (wt. %) | 98.5 | 79.65 | 79.58 | 71.2 | 66.5 | 69.21 |
| Opx (wt. %) | 0 | 17.5 | 18.21 | 23.46 | 25 | 24.5 |
| Cpx (wt. %) | 0 | 2.5 | 1.85 | 4.69 | 8.33 | 5.53 |
| Sp (wt. %) | 1.5 | 0.35 | 0.36 | 0.65 | 0.17 | 0.76 |
| <i>Major elements (wt. %)</i> | | | | | | |
| SiO₂ | 38.47 | 39.11 | 41.00 | 41.02 | 42.70 | 40.22 |
| TiO₂ | 0.04 | 0.05 | 0.05 | 0.06 | 0.06 | 0.07 |
| Al₂O₃ | 0.13 | 0.48 | 0.43 | 1.31 | 1.74 | 1.69 |
| Fe₂O₃ | 8.37 | 7.97 | 8.35 | 8.23 | 8.51 | 8.08 |
| MnO | 0.11 | 0.11 | 0.11 | 0.11 | 0.12 | 0.11 |
| MgO | 41.09 | 41.51 | 42.93 | 40.24 | 40.19 | 38.15 |
| CaO | 0.69 | 1.62 | 0.58 | 1.21 | 2.10 | 1.45 |
| Na₂O | 0.01 | 0.02 | 0.02 | 0.03 | 0.03 | 0.04 |
| K₂O | 0.02 | 0.01 | 0.02 | 0.02 | 0.02 | 0.02 |
| P₂O₅ | 0.03 | 0.02 | 0.02 | 0.02 | 0.02 | 0.02 |
| Cr₂O₃ | 0.17 | 0.19 | 0.28 | 0.41 | 0.34 | 0.28 |
| NiO | 0.30 | 0.26 | 0.29 | 0.25 | 0.25 | 0.22 |
| LOI | 10.36 | 8.86 | 5.99 | 7.37 | 4.26 | 9.52 |
| Total | 99.79 | 100.22 | 100.07 | 100.29 | 100.35 | 99.88 |
| <i>Trace elements (ppm)</i> | | | | | | |
| Sc | 6.62 | 7.66 | 9.16 | 11.3 | 14.6 | 12.0 |
| V | 18.8 | 23.4 | 28.0 | 42.5 | 64.8 | 59.5 |
| Cr | 1321 | 1448 | 2030 | 2997 | 2436 | 2089 |
| Ni | 2626 | 2262 | 2443 | 2160 | 2091 | 1953 |
| Co | 128 | 117 | 123 | 113 | 112 | 106 |
| Rb | 0.22 | 0.20 | 0.32 | 0.15 | 0.19 | 0.28 |
| Sr | 8.52 | 66.6 | 23.1 | 0.83 | 0.09 | 2.10 |
| Y | 0.02 | 0.05 | 0.04 | 0.46 | 0.57 | 0.51 |
| Zr | 3.11* | n.m. | 0.30 | 0.50 | 0.10 | 0.40 |
| Nb | 0.05 | 0.02 | 0.01 | 0.02 | 0.01 | 0.02 |
| Sb | n.m. | n.m. | n.m. | n.m. | n.m. | n.m. |
| Cs | 0.12 | 0.06 | 0.13 | 0.08 | 0.10 | 0.11 |
| Ba | 11.7 | 1.03 | 4.47 | 1.88 | 16.6 | 2.32 |
| La | 0.016 | 0.011 | 0.026 | b.d.l. | 0.025 | 0.011 |
| Ce | 0.081 | 0.025 | 0.059 | 0.023 | 0.055 | 0.023 |
| Pr | 0.005 | 0.002 | 0.004 | b.d.l. | 0.006 | 0.002 |
| Nd | 0.017 | 0.002 | 0.009 | b.d.l. | 0.016 | 0.010 |
| Sm | 0.002 | 0.001 | 0.003 | 0.011 | 0.008 | 0.005 |
| Eu | 0.003 | 0.001 | 0.002 | 0.006 | 0.006 | 0.003 |
| Gd | 0.002 | 0.006 | 0.004 | 0.029 | 0.026 | 0.018 |
| Tb | 0.000 | 0.001 | 0.001 | 0.008 | 0.008 | 0.007 |
| Dy | 0.004 | 0.011 | 0.011 | 0.069 | 0.083 | 0.069 |
| Ho | 0.001 | 0.003 | 0.003 | 0.019 | 0.023 | 0.020 |
| Er | 0.004 | 0.010 | 0.011 | 0.063 | 0.082 | 0.079 |
| Tm | 0.001 | 0.002 | 0.002 | 0.011 | 0.014 | 0.015 |
| Yb | 0.008 | 0.015 | 0.021 | 0.084 | 0.113 | 0.120 |
| Lu | 0.002 | 0.003 | 0.004 | 0.014 | 0.019 | 0.022 |
| Hf | 0.060 | 0.014 | 0.007 | 0.017 | 0.017 | 0.010 |
| Ta | 0.075 | 0.086 | 0.077 | 0.075 | 0.077 | 0.076 |
| Pb | 0.20 | b.d.l. | b.d.l. | 0.30 | 0.70 | 0.10 |
| Th | 0.012 | 0.006 | 0.008 | 0.002 | 0.010 | 0.006 |
| U | 0.090 | 0.046 | 0.014 | 0.004 | 0.033 | 0.034 |

The chemical compositions of mineral phases were measured by Cameca SX-100 four-spectrometer electron microprobe at the State Geological Institute of Dionýz Štúr in Bratislava under the operating conditions of 15 kV accelerating voltage, 20 nA focused beam current ($\phi 1-5$ nm) and 20–100 s counting time depended upon the analysed element. The standards used for calibration were: Na on albite, Si, Ca on wollastonite, K on orthoclase, Mg on forsterite, Al on Al₂O₃, Fe on fayalite, Mn on rhodonite, metallic V, Cr, and Ni, Ti on TiO₂, Sr on SrTiO₃, Nb on LiNbO₃, La on LaPO₄, Ce on CePO₄, and Ta on LiTaO₃. The mineral chemical compositions are reviewed in Tables 2–5.

Results

Petrography

The ultramafic body from Sedlice mainly consists of harzburgites (Fig. 2a from sample SE-15; SE-6b, SE-14); with lherzolites (Fig. 2b from sample SE-17b; SE-12) and dunites (samples SE-2, 3) rarely encountered. The intensively brecciated W/SW body margin was observed (Fig. 2c). The fractures of the brecciated peridotite margin are healed with secondary magnesite and calcite. The whole body is tectonically cracked into a system of blocks.

Microscopically, some ultramafics, exhibiting distinctly predominant olivine matrix (Fig. 2c), could be termed dunites. Typically mesh textured granular dunite (samples SE-2 and SE-3, Fig. 2c) is almost exclusively composed of slightly serpentinized olivine (~98–99 wt. %; from an incipient to moderate serpentinization stage) with a minimum of orthopyroxene grains (about 2–5 wt. %).

Harzburgites contain about 10–15 wt. % of orthopyroxene and less than 5 wt. % of clinopyroxene (samples SE-14 and SE-15). Orthopyroxene (1) is mostly porphyroclastic, often 1 to 2 mm in size, macroscopically visible. Harzburgite orthopyroxenes often show a strong replacement by chrysotile (Fig. 2d). Orthopyroxene porphyroclasts contain Spl inclusions (Fig. 2e,f).

Lherzolite is another rock type in the Sedlice peridotite (Fig. 2b). It usually has porphyritic texture and contains more porphyroclastic Cpx (5–10 wt. %) besides Opx (~25 wt. %) (Fig. 3a,b). Enstatite, nearly colourless with a pinkish tinge in thin sections, mainly occurs as subhedral, sometimes euhedral porphyroclastic grains up to 2 mm in size (Opx₁ in Fig. 3a,b) with a distinct exsolution lamellae system of Cpx (Fig. 3c,d). Clinopyroxene,

Table 2: The chemical compositions of pyroxenes from the Sedlice peridotite body.

| Sample | SE-6b | SE-6b | SE-6b | SE-6b | SE-6b | SE-6b | SE-6b | SE-10 | SE-10 | SE-10 |
|--------------------------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|
| Rock type | harzburgite | harzburgite | harzburgite | harzburgite | harzburgite | harzburgite | harzburgite | lherzolite | lherzolite | lherzolite |
| Mineral | Opx ₁ | Cpx ₁ | Opx ₁ | Cpx ₁ | Opx ₁ | Cpx ₁ | Opx ₁ | Opx ₁ | Cpx ₁ | Opx ₁ |
| Analysis | an25 | an26 | an28 | an27 | an29 | an30 | an24 | an8 | an7 | an13 |
| (wt. %) | | | | | | | | | | |
| SiO ₂ | 56.85 | 53.13 | 56.23 | 52.78 | 56.41 | 53.00 | 53.47 | 55.47 | 51.34 | 54.82 |
| TiO ₂ | 0.05 | 0.13 | 0.03 | 0.12 | 0.05 | 0.15 | 0.13 | 0.05 | 0.13 | 0.05 |
| Al ₂ O ₃ | 2.94 | 3.31 | 3.40 | 3.98 | 3.11 | 3.48 | 3.55 | 4.45 | 4.93 | 4.72 |
| Cr ₂ O ₃ | 0.67 | 1.03 | 0.88 | 1.33 | 0.87 | 1.38 | 1.14 | 0.93 | 1.30 | 0.97 |
| FeO | 5.71 | 2.34 | 5.66 | 2.15 | 5.76 | 1.85 | 2.35 | 5.88 | 2.35 | 5.96 |
| MnO | 0.12 | 0.08 | 0.10 | 0.08 | 0.15 | 0.09 | 0.08 | 0.16 | 0.09 | 0.13 |
| MgO | 34.04 | 17.07 | 33.41 | 16.59 | 33.70 | 16.47 | 17.48 | 32.09 | 16.01 | 32.86 |
| CaO | 0.89 | 23.39 | 0.97 | 23.41 | 0.65 | 23.75 | 22.53 | 0.92 | 24.02 | 0.95 |
| NiO | 0.11 | 0.07 | 0.09 | 0.08 | 0.13 | 0.08 | 0.03 | 0.08 | 0.06 | 0.13 |
| Na ₂ O | 0 | 0.33 | 0.01 | 0.41 | 0.02 | 0.45 | 0.35 | 0 | 0.14 | 0.01 |
| K ₂ O | 0 | 0 | 0.01 | 0.02 | 0.01 | 0.01 | 0.01 | 0.01 | 0.01 | 0.01 |
| Total | 101.38 | 100.87 | 100.78 | 100.96 | 100.85 | 100.70 | 101.12 | 100.03 | 100.38 | 100.63 |
| (a.p.f.u.) | | | | | | | | | | |
| Si ⁴⁺ | 1.935 | 1.912 | 1.927 | 1.899 | 1.931 | 1.912 | 1.917 | 1.922 | 1.863 | 1.883 |
| Al ³⁺ | 0.065 | 0.088 | 0.073 | 0.101 | 0.069 | 0.088 | 0.083 | 0.078 | 0.137 | 0.117 |
| Fe ³⁺ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| Σ | 2.000 | 2.000 | 2.000 | 2.000 | 2.000 | 2.000 | 2.000 | 2.000 | 2.000 | 2.000 |
| Al ³⁺ | 0.053 | 0.052 | 0.065 | 0.068 | 0.057 | 0.060 | 0.066 | 0.104 | 0.073 | 0.074 |
| Fe ³⁺ | 0.000 | 0.027 | 0.000 | 0.024 | 0.000 | 0.017 | 0.005 | 0.000 | 0.034 | 0.023 |
| Ti ⁴⁺ | 0.001 | 0.004 | 0.001 | 0.003 | 0.001 | 0.004 | 0.003 | 0.001 | 0.003 | 0.001 |
| Cr ³⁺ | 0.018 | 0.029 | 0.024 | 0.038 | 0.024 | 0.039 | 0.032 | 0.025 | 0.037 | 0.026 |
| Mg ²⁺ | 0.928 | 0.889 | 0.911 | 0.867 | 0.918 | 0.879 | 0.893 | 0.870 | 0.852 | 0.875 |
| Fe ²⁺ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| Mn ²⁺ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| Σ | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 |
| Mg ²⁺ | 0.798 | 0.027 | 0.796 | 0.023 | 0.802 | 0.006 | 0.041 | 0.788 | 0.014 | 0.808 |
| Fe ²⁺ | 0.163 | 0.044 | 0.162 | 0.041 | 0.165 | 0.038 | 0.066 | 0.170 | 0.037 | 0.148 |
| Mn ²⁺ | 0.003 | 0.003 | 0.003 | 0.002 | 0.004 | 0.003 | 0.002 | 0.005 | 0.003 | 0.004 |
| Ca ²⁺ | 0.033 | 0.902 | 0.036 | 0.902 | 0.024 | 0.918 | 0.865 | 0.034 | 0.934 | 0.035 |
| Na ⁺ | 0.000 | 0.023 | 0.000 | 0.029 | 0.002 | 0.031 | 0.024 | 0.000 | 0.010 | 0.001 |
| Σ | 0.997 | 0.998 | 0.997 | 0.997 | 0.996 | 0.997 | 0.999 | 0.997 | 0.998 | 0.996 |
| Wo | 1.69 | 45.81 | 1.87 | 46.29 | 1.25 | 47.37 | 43.92 | 1.83 | 46.13 | 1.88 |
| En | 89.85 | 51.71 | 89.61 | 51.36 | 90.11 | 50.44 | 52.39 | 89.02 | 51.65 | 90.19 |
| Fs | 8.46 | 2.47 | 8.52 | 2.35 | 8.63 | 2.19 | 3.69 | 9.15 | 2.22 | 7.94 |

diopside and/or augite (Cpx₁, Fig. 3b) is colourless in thin section, sometimes slightly brownish to greenish, usually with Opx exsolution lamellae. Aggregates of Opx₂ and Cpx₂ (Fig. 3e,f) with anhedral shape and missing exsolution lamellae (with exception of a few grains) represent matrix pyroxene equivalents, surrounding Opx₁ and Cpx₁ porphyroclasts. Pyroxene can be partly replaced by metamorphic amphibole-tremolite, and spinel by chromite (Fig. 2g,h). The magmatic Spl is often reddish-brownish in thin section (Fig. 3e,f).

On the basis of calculated “modal” mineral compositions (Table 1), the studied ultramafic rocks can be classified as dunites (sample SE-2, 3 in Table 1 and 4 and Fig. 4), harzburgites (samples SE-6B, 14, 15 in Table 1 and Fig. 4) to lherzolites (samples SE-12 and 17B in Table 1 and Fig. 4). The Ol-Opx-Cpx classification diagram for peridotites (Fig. 4) with plotted representative samples from the Sedlice peridotite body was constructed on the basis of calculated wt. % of mineral phases (Table 1).

Mineral abbreviations used in text, tables and figures (with exception of MgChr) are after Whitney & Evans (2010). Amp=amphibole, Cpx=clinopyroxene, Chr=chrom-

ite, Ctl=chrysotile, Fo=forsterite, Ol=olivine, Opx=orthopyroxene, Spl=spinel, Srp=serpentine group, Mag=magnetite, MgChr=magnesiochromite, Tr=tremolite.

Microprobe mineral composition and whole-rock chemical data

The major element chemical compositions of minerals are listed in Tables 2-4.

The percentages of individual components of Wo, En and Fs in the Morimoto (1988) classification diagram for Ca-Mg-Fe pyroxenes (Quad) classifies Opx as enstatite and Cpx as diopside and augite (Fig. 5, Table 2). The Mg# of pyroxenes varies from ~0.90 to 0.92 in Opx and from ~0.92 to 0.95 in Cpx (Fig. 6). Al₂O₃ contents of Opx and Cpx range from 1 to 5.58 and from 0.81 to 5.45 wt. %, respectively (Fig. 6). Orthopyroxene and clinopyroxene in lherzolites have higher Al₂O₃ contents and lower Mg#[= Mg/(Mg + Fe) atomic ratio] than those in dunites and harzburgites. The Na₂O content is very low (<0.5 wt. %) in Cpx.

The spinel-group minerals are the major accessory minerals in these rocks. According to Lindsley (1991) based on the

Table 2: Continued.

| Sample | SE-10 | SE-10 | SE-10 | SE-10 | SE-15 | SE-15 | SE-15 | SE-15 | SE-15 | SE-15 | SE-15 |
|--------------------------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|------------------|
| Rock type | lherzolite | lherzolite | lherzolite | lherzolite | harzburgite | harzburgite | harzburgite | harzburgite | harzburgite | harzburgite | harzburgite |
| Mineral | Cpx ₁ | Opx ₁ | Cpx ₁ | Cpx ₂ | Opx ₁ | Opx ₂ | Opx ₁ | Opx ₂ | Cpx ₂ | Cpx ₂ | Cpx ₂ |
| Analysis | an12 | an14 | an15 | an11 | an4 | an8 | an21 | an23 | an3 | an6 | an16 |
| (wt. %) | | | | | | | | | | | |
| SiO ₂ | 51.24 | 55.15 | 51.67 | 52.52 | 56.92 | 57.15 | 57.46 | 57.40 | 53.56 | 53.83 | 54.38 |
| TiO ₂ | 0.15 | 0.02 | 0.11 | 0.10 | 0.02 | 0.01 | 0.01 | 0.03 | 0.02 | 0.00 | 0.01 |
| Al ₂ O ₃ | 5.15 | 4.84 | 4.77 | 4.04 | 1.36 | 1.02 | 1.14 | 1.23 | 1.09 | 0.81 | 1.13 |
| Cr ₂ O ₃ | 1.25 | 0.89 | 1.05 | 0.80 | 0.56 | 0.33 | 0.46 | 0.51 | 0.53 | 0.33 | 0.73 |
| FeO | 2.13 | 6.26 | 2.49 | 2.52 | 5.53 | 5.32 | 5.20 | 5.50 | 2.06 | 1.89 | 1.79 |
| MnO | 0.08 | 0.09 | 0.13 | 0.11 | 0.16 | 0.14 | 0.13 | 0.16 | 0.08 | 0.05 | 0.07 |
| MgO | 16.11 | 32.62 | 16.56 | 17.07 | 34.38 | 34.63 | 34.75 | 34.09 | 18.13 | 17.92 | 18.15 |
| CaO | 24.03 | 0.54 | 23.51 | 23.26 | 0.87 | 0.77 | 0.75 | 0.90 | 23.64 | 24.11 | 24.03 |
| NiO | 0.00 | 0.12 | 0.05 | 0.04 | 0.09 | 0.10 | 0.06 | 0.06 | 0.05 | 0.09 | 0.04 |
| Na ₂ O | 0.10 | 0.03 | 0.07 | 0.11 | 0 | 0 | 0.01 | 0 | 0.07 | 0.07 | 0.07 |
| K ₂ O | 0 | 0 | 0 | 0 | 0 | 0 | 0.01 | 0.01 | 0.01 | 0.01 | 0 |
| Total | 100.24 | 100.57 | 100.42 | 100.58 | 99.89 | 99.46 | 99.99 | 99.89 | 99.22 | 99.12 | 100.40 |
| (a.p.f.u.) | | | | | | | | | | | |
| Si ⁴⁺ | 1.860 | 1.898 | 1.871 | 1.896 | 1.963 | 1.976 | 1.976 | 1.982 | 1.954 | 1.967 | 1.962 |
| Al ³⁺ | 0.140 | 0.102 | 0.129 | 0.104 | 0.037 | 0.024 | 0.024 | 0.018 | 0.046 | 0.033 | 0.038 |
| Fe ³⁺ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| Σ | 2.000 | 2.000 | 2.000 | 2.000 | 2.000 | 2.000 | 2.000 | 2.000 | 2.000 | 2.000 | 2.000 |
| Al ³⁺ | 0.080 | 0.094 | 0.075 | 0.068 | 0.018 | 0.017 | 0.022 | 0.032 | 0.001 | 0.002 | 0.011 |
| Fe ³⁺ | 0.024 | 0.000 | 0.027 | 0.018 | 0.008 | 0.004 | 0.000 | 0.000 | 0.037 | 0.031 | 0.013 |
| Ti ⁴⁺ | 0.004 | 0.001 | 0.003 | 0.003 | 0.001 | 0.000 | 0.000 | 0.001 | 0.001 | 0.000 | 0.000 |
| Cr ³⁺ | 0.036 | 0.024 | 0.030 | 0.023 | 0.015 | 0.009 | 0.013 | 0.014 | 0.015 | 0.010 | 0.021 |
| Mg ²⁺ | 0.857 | 0.881 | 0.866 | 0.889 | 0.958 | 0.970 | 0.965 | 0.954 | 0.946 | 0.957 | 0.955 |
| Fe ²⁺ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| Mn ²⁺ | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 | 0.000 |
| Σ | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 | 1.000 |
| Mg ²⁺ | 0.015 | 0.792 | 0.028 | 0.030 | 0.809 | 0.815 | 0.816 | 0.801 | 0.040 | 0.020 | 0.021 |
| Fe ²⁺ | 0.041 | 0.180 | 0.049 | 0.058 | 0.151 | 0.150 | 0.150 | 0.159 | 0.026 | 0.027 | 0.041 |
| Mn ²⁺ | 0.003 | 0.003 | 0.004 | 0.003 | 0.005 | 0.004 | 0.004 | 0.005 | 0.002 | 0.002 | 0.002 |
| Ca ²⁺ | 0.934 | 0.020 | 0.912 | 0.900 | 0.032 | 0.029 | 0.028 | 0.033 | 0.924 | 0.944 | 0.929 |
| Na ⁺ | 0.007 | 0.002 | 0.005 | 0.008 | 0.000 | 0.000 | 0.001 | 0.000 | 0.005 | 0.005 | 0.005 |
| Σ | 1.000 | 0.996 | 0.998 | 0.999 | 0.998 | 0.997 | 0.998 | 0.998 | 0.998 | 0.997 | 0.999 |
| Wo | 46.05 | 1.06 | 44.66 | 44.55 | 1.65 | 1.46 | 1.42 | 1.72 | 45.50 | 46.84 | 46.43 |
| En | 51.54 | 89.33 | 52.49 | 52.14 | 90.59 | 90.91 | 90.94 | 90.12 | 53.09 | 51.75 | 51.42 |
| Fs | 2.41 | 9.61 | 2.86 | 3.31 | 7.76 | 7.63 | 7.64 | 8.16 | 1.41 | 1.41 | 2.15 |

mineral chemical compositions (Table 3), they are classified as spinel, magnesiochromite and chromite end-members (Fig. 7a). The harzburgites contain only magnesiochromite and chromite, whereas lherzolites also preserve the spinel end member. Chromite grains are round-shaped or euhedral, sometimes weakly cataclastic. They are accumulated along mineral grain boundaries, in serpentine veins in the form of mineral inclusions or dispersed as small and/or large grain clusters. Magnetite is the youngest of the spinel-group minerals. It occurs in the form of large patches around primary reddish spinel, or it kneads the cataclastic chromite in the form of veins. A single grain of andradite garnet was found in the more altered part of harzburgite.

Spinel compositions vary widely between samples. Spinel in harzburgites is represented by rounded, usually euhedral to subhedral chromite and vermicular magnesiochromite. It is systematically higher in Cr#[= Cr/(Cr+Al) atomic ratio; Cr# = 0.58–0.85] and lower in Mg#[= Mg/(Mg+Fe²⁺) atomic ratio; Mg# = 0.37–0.60] than that in lherzolites, represented by reddish spinel (Cr# = 0.14–0.37; Mg# = 0.66–0.79; Fig. 7a). The TiO₂ content is low (<0.22 wt. %; Fig. 7b) in all cases, only the sample SE-14 represented by harzburgite has signif-

icantly higher TiO₂ contents in the range from 0.18 to 0.21 compared to other samples. Cr# in Spl/Fo in Ol diagram (Fig. 8a) and Cr₂O₃/Al₂O₃ Spl diagram (Fig. 8b) plot most of the samples in the Mantle-array. The only exception is sample SE-3 containing Spl with an increased Cr#, which is compatible with the dunite as the hosting rock.

Olivine is the most representative rock-forming mineral with forsterite content ranging from 89.0 to 91.4 in lherzolites, and from 90.0 to 92.0 in harzburgites. The NiO content of olivine ranges from 0.31 to 0.47 wt. % in harzburgites, and from 0.36 to 0.51 wt. % in lherzolites (Table 4). The relationship between the Fo content of olivine and Cr# of spinel shows that all lherzolites and harzburgites plot into the Ol-Spl mantle array (OSMA), a residual mantle trend of spinel peridotite (Arai 1987, 1994; Fig. 8a). Olivine composition is in correlation with spinel; spinel Cr# increases and spinel Mg# decreases as olivine Fo increases in the OSMA (Choi et al. 2008).

The Primitive Mantle (PM) normalized rare earth element concentrations from the whole-rock analyses are plotted in Fig. 9. They show a decrease in REE as a whole, but a relative increase in LREE, particularly Ce and La, in comparison with HREE. Three to four samples exhibit a positive Eu anomaly.

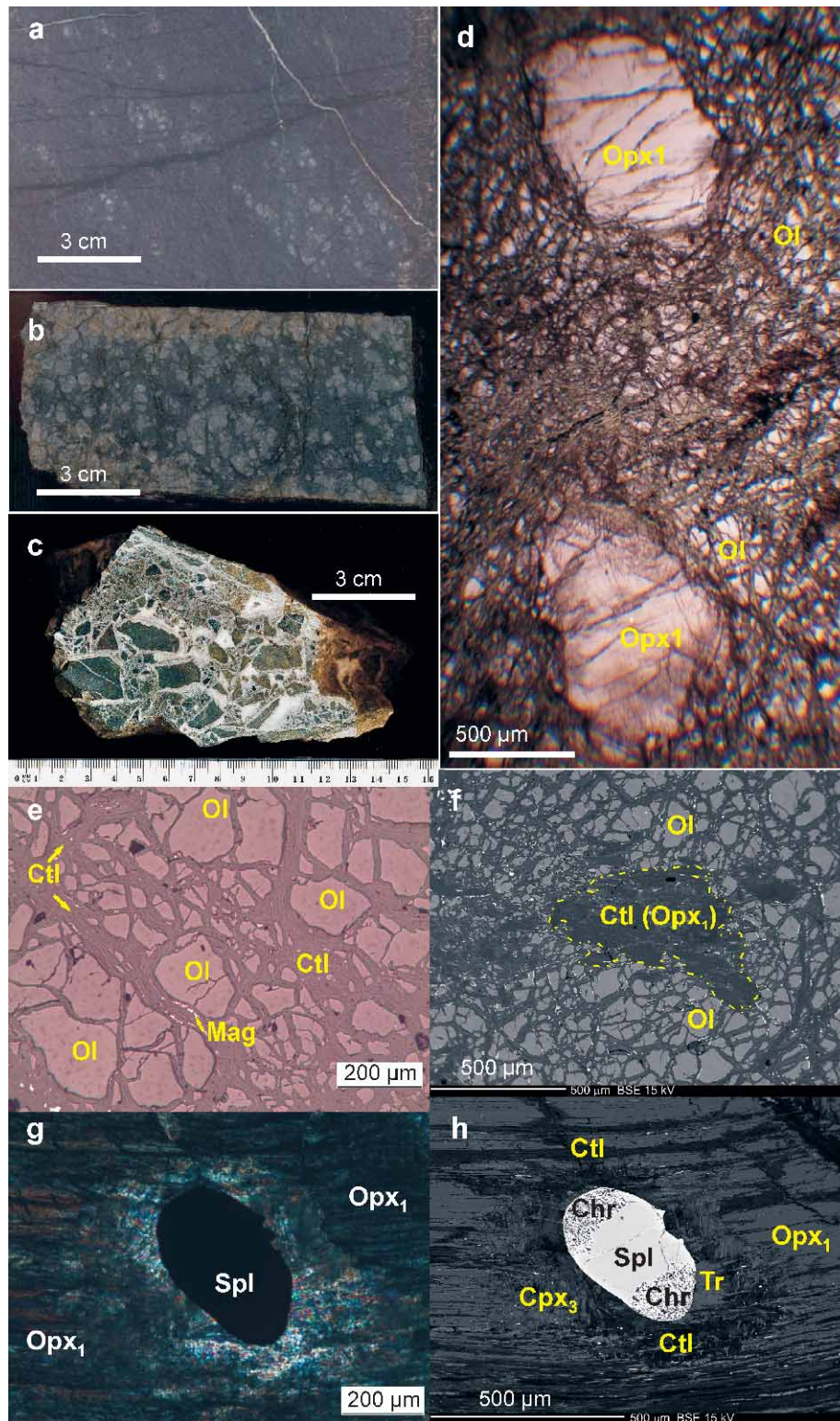


Fig. 2. **a** — Serpentinized harzburgite (sample SE-15) from Sedlice peridotite with preserved pyroxene porphyroclasts; **b** — Slightly serpentinized lherzolite from Sedlice peridotite (sample SE-17b); **c** — Tectonic breccia composed of harzburgite angular fragments cemented by carbonates — calcite and magnesite; **d** — Photomicrograph (cross-polarized light) of harzburgite (sample SE-15) with Opx₁ porphyroclasts in olivine-rich matrix; **e** — Photomicrograph (reflected light) of granular dunite (sample SE-3) from Sedlice locality with typical mesh textured serpentine minerals forming veins surrounding the relic center of olivine grains; **f** — BSE image of serpentinized dunite (sample SE-3) with outlined boundary of rare orthopyroxene (1); **g** — Photomicrograph (cross-polarized light); **h** — BSE image of spinel rim dissolution-precipitation replacement (Putnis 2009) by metamorphic chromite in host orthopyroxene (1) porphyroclast in Sedlice serpentinized harzburgite (sample SE-6b). Orthopyroxene partly replaced by metamorphic Srp and Tr.

Table 3: The chemical compositions of spinel group minerals from the Sedlice peridotite body.

| Sample Rock type Mineral Ana No. | SE-3 | | SE-3 | | SE-6b | | SE-6b | | SE-12 | | SE-12 | | SE-14 | | SE-14 | | SE-15 | | SE-15 | | SE-17b | | SE-17b | | |
|---|--------|-------|--------|-------|--------|--------|--------|--------|-------|-------|-------|-------|--------|--------|--------|-------|--------|--------|-------|--------|--------|--------|--------|-------|-------|
| | dun | Chr | dun | Chr | harz | Spl | harz | Spl | lherz | Spl | harz | MgChr | harz | MgChr | harz | Chr | harz | Chr | harz | Spl | lherz | Spl | lherz | Spl | lherz |
| (wt. %) | | | | | | | | | | | | | | | | | | | | | | | | | |
| TiO ₂ | 0.05 | 0.06 | 0.05 | 0.07 | 0.08 | 0.05 | 0.07 | 0.08 | 0.02 | 0.02 | 0.06 | 0.21 | 0.19 | 0.20 | 0.04 | 0.03 | 0.05 | 0.05 | 0.08 | 0.07 | 0.07 | 0.07 | 0.07 | 0.05 | 0.05 |
| Al ₂ O ₃ | 7.40 | 7.25 | 7.55 | 36.32 | 36.59 | 38.74 | 36.59 | 38.74 | 46.65 | 55.08 | 44.45 | 21.03 | 21.14 | 21.20 | 19.03 | 21.52 | 22.08 | 22.08 | 49.39 | 50.38 | 50.38 | 50.38 | 51.67 | 51.67 | |
| V ₂ O ₅ | 0.21 | 0.29 | 0.34 | 0.12 | 0.11 | 0.12 | 0.11 | 0.12 | 0.12 | 0.12 | 0.12 | 0.21 | 0.22 | 0.22 | 0.24 | 0.26 | 0.19 | 0.19 | 0.16 | 0.13 | 0.13 | 0.13 | 0.13 | 0.13 | |
| Cr ₂ O ₃ | 62.53 | 61.50 | 62.24 | 31.45 | 32.66 | 31.10 | 32.66 | 31.10 | 21.15 | 12.93 | 23.42 | 47.60 | 48.15 | 47.97 | 50.27 | 45.82 | 45.86 | 45.86 | 19.54 | 18.90 | 18.90 | 18.90 | 16.64 | 16.64 | |
| FeO | 21.49 | 22.11 | 22.17 | 14.99 | 14.78 | 14.64 | 14.78 | 14.64 | 12.65 | 10.86 | 13.17 | 17.52 | 17.70 | 17.65 | 18.40 | 20.80 | 21.17 | 21.17 | 11.18 | 11.44 | 11.44 | 11.44 | 12.03 | 12.03 | |
| MnO | 0.56 | 0.58 | 0.58 | 0.25 | 0.21 | 0.22 | 0.21 | 0.22 | 0.19 | 0.16 | 0.18 | 0.37 | 0.37 | 0.41 | 0.39 | 0.44 | 0.44 | 0.44 | 0.16 | 0.17 | 0.17 | 0.17 | 0.17 | 0.17 | |
| MgO | 7.99 | 7.38 | 7.73 | 15.45 | 15.48 | 15.89 | 15.48 | 15.89 | 17.64 | 19.39 | 17.74 | 12.64 | 12.46 | 12.33 | 11.94 | 10.27 | 9.98 | 9.98 | 18.49 | 18.81 | 18.81 | 18.81 | 18.66 | 18.66 | |
| NiO | 0 | 0.071 | 0 | 0.15 | 0.14 | 0.15 | 0.14 | 0.15 | 0.24 | 0.30 | 0.22 | 0.05 | 0.10 | 0.06 | 0.06 | 0 | 0 | 0 | 0.20 | 0.27 | 0.27 | 0.27 | 0.15 | 0.15 | |
| ZnO | 0.21 | 0.14 | 0.24 | 0.14 | 0.19 | 0.14 | 0.19 | 0.14 | 0.21 | 0.24 | 0.21 | 0.10 | 0.08 | 0.19 | 0.21 | 0.30 | 0.43 | 0.43 | 0.25 | 0.18 | 0.18 | 0.18 | 0.23 | 0.23 | |
| Total | 100.44 | 99.39 | 100.91 | 98.92 | 100.24 | 101.06 | 100.24 | 101.06 | 98.91 | 99.10 | 99.56 | 99.73 | 100.43 | 100.21 | 100.57 | 99.46 | 100.20 | 100.20 | 99.43 | 100.35 | 100.35 | 100.35 | 99.73 | 99.73 | |
| (a.p.f.u.) | | | | | | | | | | | | | | | | | | | | | | | | | |
| Ti ⁴⁺ | 0.001 | 0.001 | 0.001 | 0.002 | 0.002 | 0.001 | 0.002 | 0.001 | 0.000 | 0.000 | 0.001 | 0.005 | 0.004 | 0.005 | 0.001 | 0.001 | 0.001 | 0.001 | 0.002 | 0.001 | 0.001 | 0.001 | 0.001 | 0.001 | |
| Al ³⁺ | 0.294 | 0.293 | 0.299 | 1.236 | 1.232 | 1.283 | 1.232 | 1.283 | 1.510 | 1.710 | 1.442 | 0.769 | 0.769 | 0.773 | 0.700 | 0.800 | 0.815 | 0.815 | 1.570 | 1.582 | 1.582 | 1.582 | 1.623 | 1.623 | |
| V ³⁺ | 0.006 | 0.008 | 0.009 | 0.003 | 0.003 | 0.003 | 0.003 | 0.003 | 0.004 | 0.003 | 0.003 | 0.003 | 0.005 | 0.005 | 0.006 | 0.007 | 0.005 | 0.005 | 0.003 | 0.003 | 0.003 | 0.003 | 0.003 | 0.003 | |
| Cr ³⁺ | 1.668 | 1.665 | 1.655 | 0.718 | 0.738 | 0.691 | 0.738 | 0.691 | 0.459 | 0.269 | 0.510 | 1.168 | 1.175 | 1.174 | 1.240 | 1.142 | 1.135 | 1.135 | 0.417 | 0.398 | 0.398 | 0.398 | 0.351 | 0.351 | |
| Fe ³⁺ | 0.036 | 0.041 | 0.044 | 0.044 | 0.028 | 0.024 | 0.028 | 0.024 | 0.031 | 0.021 | 0.047 | 0.059 | 0.052 | 0.048 | 0.059 | 0.058 | 0.048 | 0.048 | 0.012 | 0.019 | 0.019 | 0.019 | 0.025 | 0.025 | |
| Fe ²⁺ | 0.570 | 0.592 | 0.579 | 0.318 | 0.325 | 0.320 | 0.325 | 0.320 | 0.260 | 0.218 | 0.256 | 0.396 | 0.405 | 0.409 | 0.421 | 0.491 | 0.506 | 0.506 | 0.236 | 0.236 | 0.236 | 0.236 | 0.243 | 0.243 | |
| Mn ²⁺ | 0.016 | 0.017 | 0.017 | 0.006 | 0.005 | 0.005 | 0.005 | 0.005 | 0.004 | 0.004 | 0.004 | 0.010 | 0.010 | 0.011 | 0.010 | 0.012 | 0.012 | 0.012 | 0.004 | 0.004 | 0.004 | 0.004 | 0.004 | 0.004 | |
| Mg ²⁺ | 0.402 | 0.377 | 0.388 | 0.665 | 0.659 | 0.666 | 0.659 | 0.666 | 0.722 | 0.761 | 0.728 | 0.585 | 0.573 | 0.569 | 0.555 | 0.483 | 0.466 | 0.466 | 0.743 | 0.747 | 0.747 | 0.747 | 0.741 | 0.741 | |
| Ni ²⁺ | 0.000 | 0.002 | 0.000 | 0.003 | 0.003 | 0.003 | 0.003 | 0.003 | 0.005 | 0.006 | 0.005 | 0.001 | 0.002 | 0.001 | 0.002 | 0.000 | 0.000 | 0.000 | 0.004 | 0.006 | 0.006 | 0.006 | 0.003 | 0.003 | |
| Zn ²⁺ | 0.005 | 0.004 | 0.006 | 0.003 | 0.004 | 0.003 | 0.004 | 0.003 | 0.004 | 0.005 | 0.004 | 0.002 | 0.002 | 0.004 | 0.005 | 0.007 | 0.010 | 0.010 | 0.005 | 0.004 | 0.004 | 0.004 | 0.004 | 0.004 | |
| Total | 2.999 | 3.000 | 2.999 | 2.998 | 3.000 | 3.000 | 3.000 | 3.000 | 3.000 | 2.997 | 2.999 | 2.999 | 2.998 | 2.999 | 2.999 | 2.999 | 2.998 | 2.998 | 3.000 | 2.999 | 2.999 | 2.999 | 2.999 | 2.999 | |
| Mg# | 0.414 | 0.389 | 0.401 | 0.677 | 0.670 | 0.676 | 0.670 | 0.676 | 0.735 | 0.777 | 0.740 | 0.596 | 0.586 | 0.582 | 0.569 | 0.496 | 0.479 | 0.479 | 0.756 | 0.760 | 0.760 | 0.760 | 0.753 | 0.753 | |
| Cr# | 0.850 | 0.851 | 0.847 | 0.367 | 0.374 | 0.350 | 0.374 | 0.350 | 0.233 | 0.136 | 0.261 | 0.603 | 0.604 | 0.603 | 0.639 | 0.588 | 0.582 | 0.582 | 0.210 | 0.201 | 0.201 | 0.201 | 0.178 | 0.178 | |

Discussion

The dunites represent the most refractory residual mantle rocks observed in the Sedlice peridotite body in association with prevailing harzburgites and exclusively with lherzolites. The studied rocks from this body are characterized by a wide range of Cr# (from 0.14 to 0.85) in Spl. The relationship between Cr# of spinel and Fo of olivine indicates that studied rocks are within the OSMA (Arai 1987, 1994) and represent residual equivalents of a mantle peridotite which compositionally reveal a wide range of melt extraction derived by partial melting of a more fertile mantle peridotite (Fig. 8a,b). The Fo content of olivine and Cr# of spinel do not change during subsolidus recrystallization (Ozawa 1988; Arai 1994). The Cr# of spinel-group minerals is progressively increasing with partial melting degree, which, on the other hand, reduces the original Al contents in Spl and Opx, and the host rock (Dick & Bullen 1984; Arai 1994; Ohara & Ishi 1998). Therefore such a wide range of Cr# in spinel in our rocks is good indicator of a wide range of degrees of partial melting. Based on the cited concepts, the chemical compositions of chromian spinels (Cr-Spl) are plotted in various discrimination diagrams. The spinel data for lherzolite (SE-12, SE-17b) and harzburgite (SE-6b, SE-14, SE-15) are plotted within the mantle array on an Al₂O₃/Cr₂O₃ diagram (Fig. 8b). In the diagram Mg#/Cr# (Fig. 7a) the most Cr-spinels from lherzolite (i.e. plot of the lowest Cr# values; samples SE-12 and SE-17b) indicate its relatively undepleted nature. On the other hand, the Cr-spinel data of refractory harzburgite (samples SE-14 and SE-15) indicates depleted character. The highest depletion of the mantle rocks is inferred in an arc tectonic setting and/or in a supra-subduction zone environment (e.g. Arai 1994; Choi et al. 2008). The Cr# of spinel in abyssal peridotites collected from mid-ocean ridges is less than 0.6 (Tamura & Arai 2006). Our samples have mostly comparable (SE-15, SE-14), but also slightly higher (SE-3) values (Figs. 7, 8). The TiO₂ content is extremely low in arc/back-arc magma, intermediate in mature mid-oceanic ridge magma and high in intra-plate magma (Arai 1992). The spinel TiO₂ contents plotted in Fig. 7b record low concentrations in all samples. The diagram Mg# versus Al₂O₃ content in pyroxenes

(Fig. 6) shows differences in composition between lherzolite and harzburgite most likely due to variable degrees of their depletion.

Trace rare earth element concentrations from whole-rock analyses are plotted in Fig. 9 and exhibit a depletion trend in harzburgites, corresponding to their refractory origin in

relationship to more fertile lherzolites. The basic trend in PM normalized REE patterns (Fig. 9a) indicates a depleted rock-suite, however an increase in LREE to HREE is similar to the Dobšiná meta-harzburgites (Putiš et al. 2012), likely indicating an influence of metamorphic fluids during serpentinization. The whole-rock positive Eu anomaly suggests

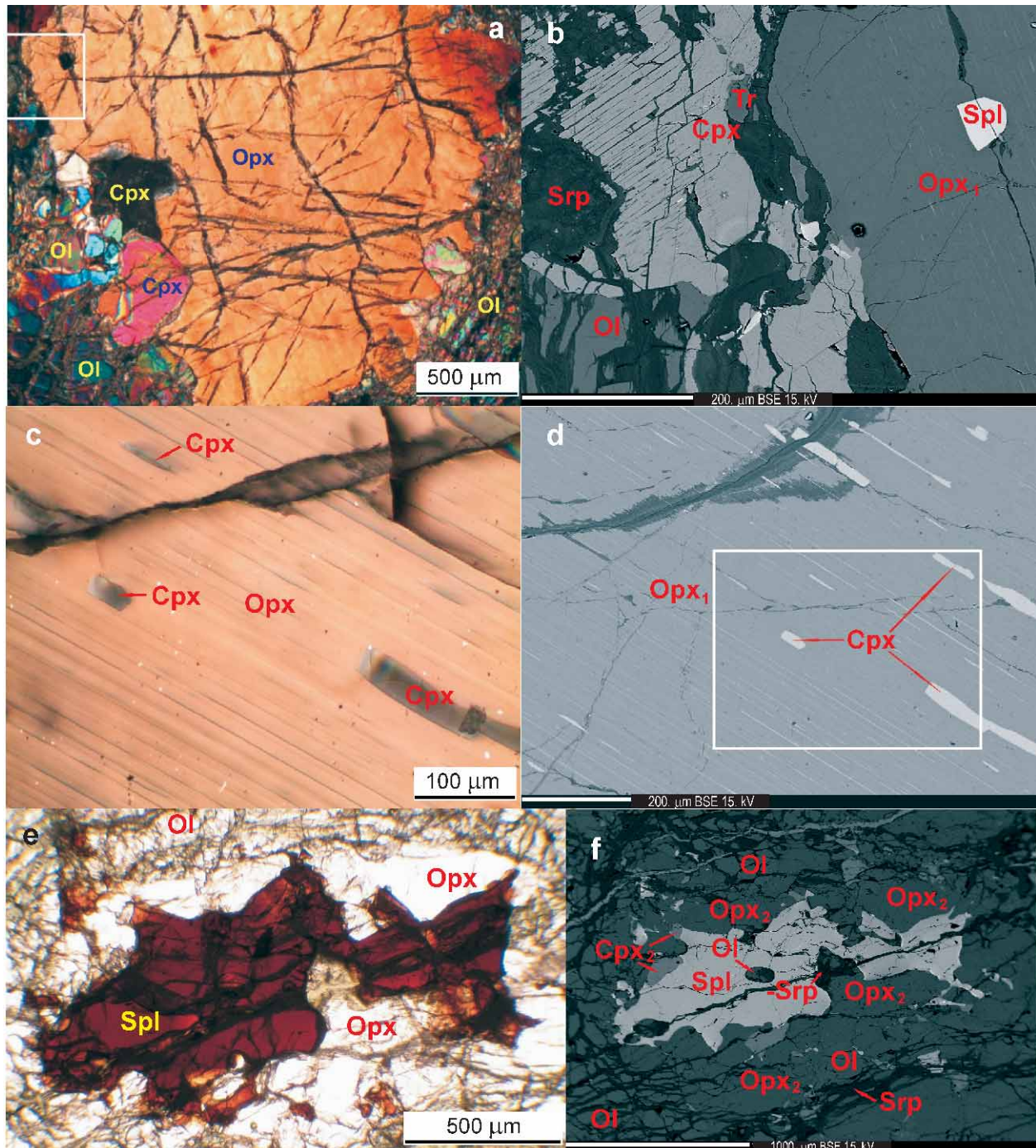


Fig. 3. Photomicrographs of a slightly serpentinized Spl lherzolite (samples SE-12) from Sedlice peridotite. **a** — Photomicrograph (cross-polarized light) of porphyroclastic Opx_1 with Spl inclusion (in the frame); **b** — BSE image of Spl inclusion in Opx_1 (from image in **a**); **c** — Photomicrograph (cross-polarized light) of exsolved clinopyroxene lamellae in host Opx_1 (a BSE image detail in **d** frame); **d** — BSE image detail of **c** in the frame; **e** — Photomicrograph (plain-polarized light) of brown-reddish spinel in Opx_2 - Cpx_2 aggregate; **f** — BSE image of spinel from **e**.

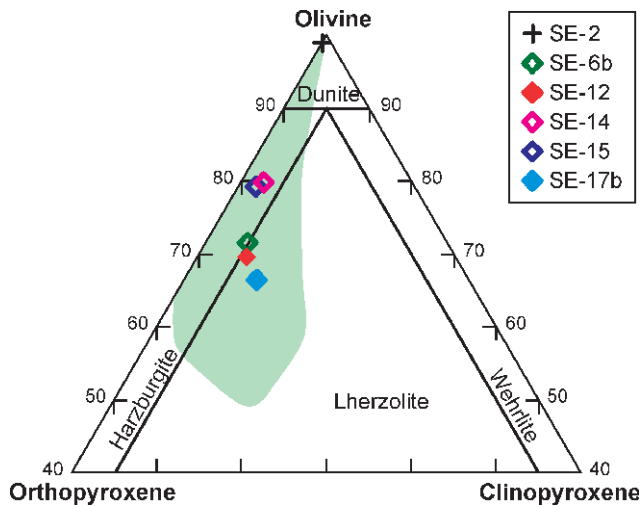


Fig. 4. Classification diagram for peridotites with representative samples from Sedlice peridotite body plotted and calculated on the basis of primary mineral phases and whole-rock chemical compositions (mineral quantities are calculated in wt. %; Table 1). The pale green area represents the most common compositions of peridotites in the upper part of the Earth's mantle (partly adapted from Bodinier & Godard (2007)).

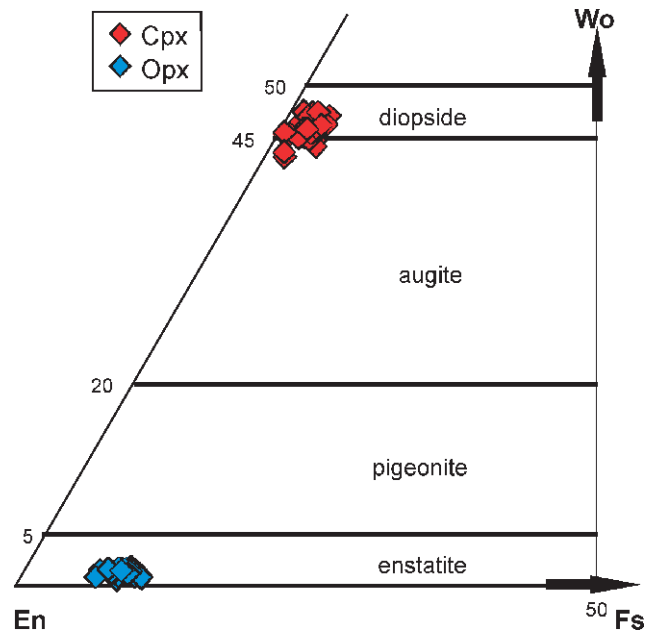


Fig. 5. Pyroxenes from Sedlice peridotite in the Morimoto (1988) classification.

Table 4: Olivine chemical compositions from the Sedlice peridotite body.

| Sample | SE-3 | SE-6b | SE-14 | SE-15 | SE-12 | SE-17b |
|----------------------|--------|-------|--------|--------|--------|--------|
| Rock type | dun | harz | harz | harz | lherz | lherz |
| # of analyses | 6 | 16 | 6 | 7 | 5 | 4 |
| (wt. %) | | | | | | |
| SiO ₂ | 41.37 | 40.98 | 41.61 | 41.38 | 40.98 | 41.38 |
| FeO _(tot) | 8.28 | 8.82 | 8.49 | 8.03 | 9.01 | 9.05 |
| MnO | 0.12 | 0.13 | 0.11 | 0.12 | 0.13 | 0.14 |
| MgO | 50.33 | 49.42 | 50.22 | 50.04 | 49.67 | 49.27 |
| CaO | 0.01 | 0.03 | 0.01 | 0.02 | 0.03 | 0.25 |
| NiO | 0.40 | 0.41 | 0.42 | 0.39 | 0.42 | 0.42 |
| Total | 100.56 | 99.87 | 100.92 | 100.06 | 100.33 | 100.60 |
| Mg# | 0.92 | 0.91 | 0.91 | 0.92 | 0.91 | 0.91 |
| Fo | 91.18 | 90.50 | 90.95 | 91.37 | 90.35 | 89.98 |

Table 5: The chemical compositions of metamorphic minerals from the Sedlice peridotite body.

| Sample | SE-3 | SE-6b | SE-6b | SE-10 | SE-6b | SE-6b | SE-15 | SE-17b |
|--------------------------------|-------|-------|-------|-------|------------------|------------------|-------|--------|
| Rock type | dun | harz | harz | lherz | harz | harz | harz | lherz |
| Mineral | Srp | Srp | Srp | Srp | Cpx ₃ | Cpx ₃ | Tr | Tr |
| Ana No. | 5 | 11 | 73 | 10 | 32 | 34 | 19 | 9 |
| (wt. %) | | | | | | | | |
| SiO ₂ | 43.49 | 42.66 | 36.83 | 40.27 | 54.78 | 55.09 | 57.23 | 54.78 |
| TiO ₂ | 0 | 0 | 0.03 | 0 | 0.03 | 0.04 | 0.01 | 0.19 |
| Al ₂ O ₃ | 0.06 | 1.55 | 1.96 | 0.17 | 1.24 | 0.92 | 1.22 | 3.92 |
| Cr ₂ O ₃ | 0.01 | 0.64 | 0.83 | 0.00 | 0.35 | 0.27 | 0.50 | 0.77 |
| FeO | 1.63 | 3.20 | 6.35 | 6.66 | 2.01 | 1.57 | 1.95 | 2.18 |
| MnO | 0.03 | 0.08 | 0.19 | 0.25 | 0.12 | 0.13 | 0.04 | 0.04 |
| MgO | 42.35 | 37.89 | 36.23 | 35.90 | 21.29 | 20.38 | 23.29 | 21.90 |
| CaO | 0.02 | 0.11 | 0.06 | 0.23 | 19.08 | 22.12 | 13.50 | 13.28 |
| NiO | 0.16 | 0.02 | 0.10 | 0.36 | 0.02 | 0.15 | 0.11 | 0.08 |
| Na ₂ O | 0 | 0.09 | 0 | 0.01 | 0.55 | 0.30 | 0.13 | 0.64 |
| K ₂ O | 0.02 | 0.06 | 0.02 | 0.04 | 0.18 | 0.11 | 0.01 | 0.03 |
| Cl | 0.03 | 0.05 | 0.46 | 0.13 | 0 | 0 | 0.04 | 0.05 |
| Total | 87.80 | 86.36 | 83.06 | 84.01 | 99.64 | 101.07 | 98.04 | 97.85 |

substitution of Eu²⁺ for Ca²⁺, likely in tremolite (in the B site) or rare carbonates. The trace element mobility is also documented in spider diagram (Fig. 9b) showing a relative increase in Cs, Ba, U, Ta, Ce, Sr, Zr, Hf and Ti. The main difference between the less and more depleted rocks exhibit HREE and Ti, which are increased in (less depleted) lherzolites.

From this point of view, the harzburgites from the Sedlice peridotite body could be genetically bound to an abyssal mantle peridotite (e.g. Bodinier & Godard 2007). Some of the Spl analyses at the boundary of the mantle array could indicate either a higher melting degree, or a later influence of metamorphic fluids, or both.

The studied peridotite body exhibits features of low-temperature metamorphic overprint. Metamorphic mineral assemblage contains chrysotile, tremolite, andradite garnet, Cr-spinel to chromite and magnetite, an increase of fayalite component in olivine, and rare carbonate. This might be related to serpentinization and a weak rodingitization in an accretionary wedge, resembling the Meliatic Bôrka Nappe serpentinized and rodingitized harzburgites (Putiš et al. 2012; Li et al. 2014). Because of practically missing high-pressure metamorphic overprint, characteristic for the Meliatic Bôrka blueschist-bearing nappe (with antigorite-clinopyroxene-pargasite-bearing meta-harzburgites, Putiš et al. 2012), this body could indicate an obduction process and incorporation into an accretionary wedge due to closure of the Neotethyan Meliatic (Triassic-Jurassic) oceanic back-arc basin in the Late Jurassic (Dallmeyer et al. 1996; Mock et al. 1998; Faryad et al. 2005; Putiš et al. 2011).

The Sedlice body could be an olistolith, which slid into the Paleogene sediments from the Bôrka Nappe, overlying the Central Western Carpathians

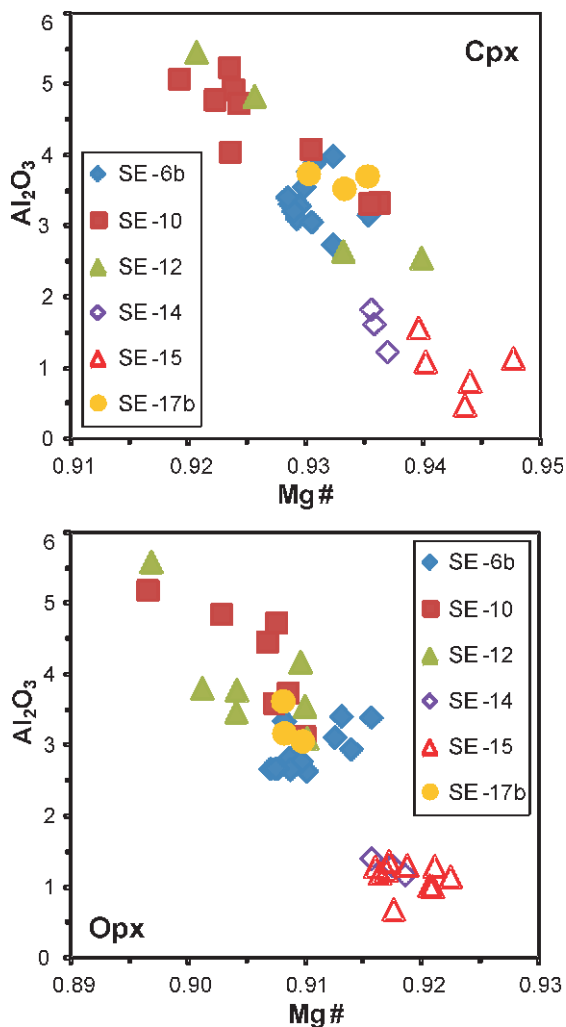


Fig. 6. Plot of $Mg\# [= Mg/(Mg + Fe)$ atomic ratio] versus Al_2O_3 contents (wt. %) in monoclinic and rhombic pyroxenes from Sedlice peridotite.

orogenic wedge since the Late Cretaceous. Alternatively it could be a younger tectonic protrusion body from the underlying basement units, most likely from the Meliatic Bôrka Nappe, which originated after covering by the Paleogene basal and younger flysch sediments. Nevertheless, the surface occurrences of the metaperidotites of the Bôrka Nappe are not so far away — along the Gemeric/Veporic Late Cretaceous tectonic boundary, to the south (SW) of the Central-Carpathian Paleogene Basin (Fig. 1). The brecciation of the body margin healed by magnesite and calcite postdates pervasive and relatively higher-temperature serpentinization most likely in an accretionary wedge. Therefore the brecciation might reflect an interaction of CO_2 -rich water with the body margin within the sedimentary basin.

Conclusions

The spinel peridotite from Sedlice comprises mainly depleted harzburgites accompanied by lherzolites and dunites, which

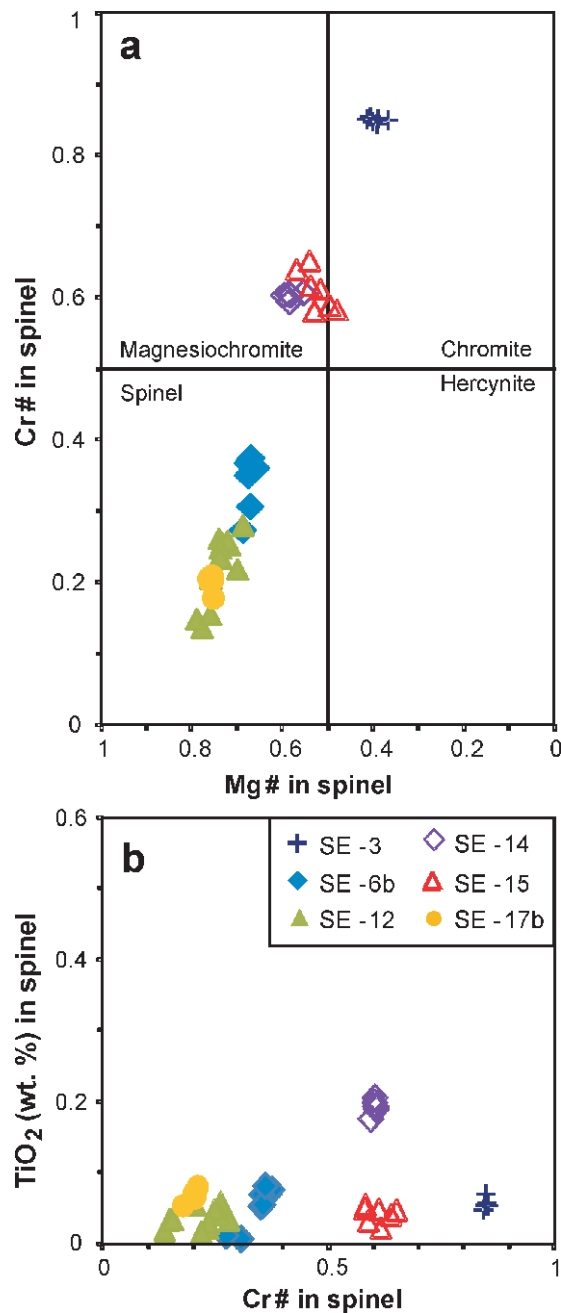


Fig. 7. Spinel compositional variations in the peridotite body from Sedlice. **a** — Relationship between $Cr\# [= Cr/(Cr + Al)$ atomic ratio] and TiO_2 content; **b** — Relationship between $Mg\# [= Mg/(Mg + Fe^{2+})$ atomic ratio] and $Cr\#$. The individual spinel-group end-member mineral terminology is after Lindsley (1991).

formed in a spinel stability field mantle environment. The studied samples point to a wide range of depletion. By composition, the lherzolites are close to fertile spinel lherzolites. The harzburgites represent depleted mantle rocks or residual equivalents formed by various degree of partial melting of more fertile lherzolites. The depletion is recorded in their refractory lithology and mineral chemical compositions. Cumulates, the expected derivatives formed by magmatic differentiation or a higher degree partial melting were not found in this body.

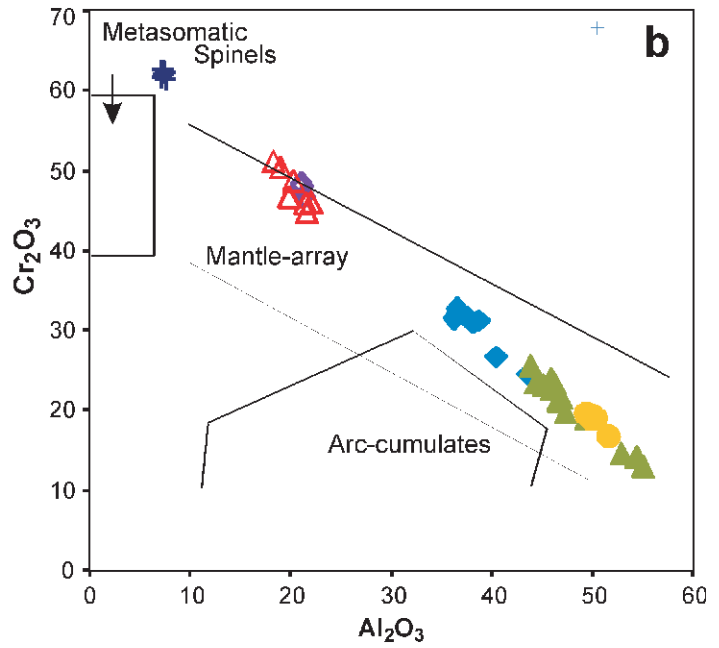
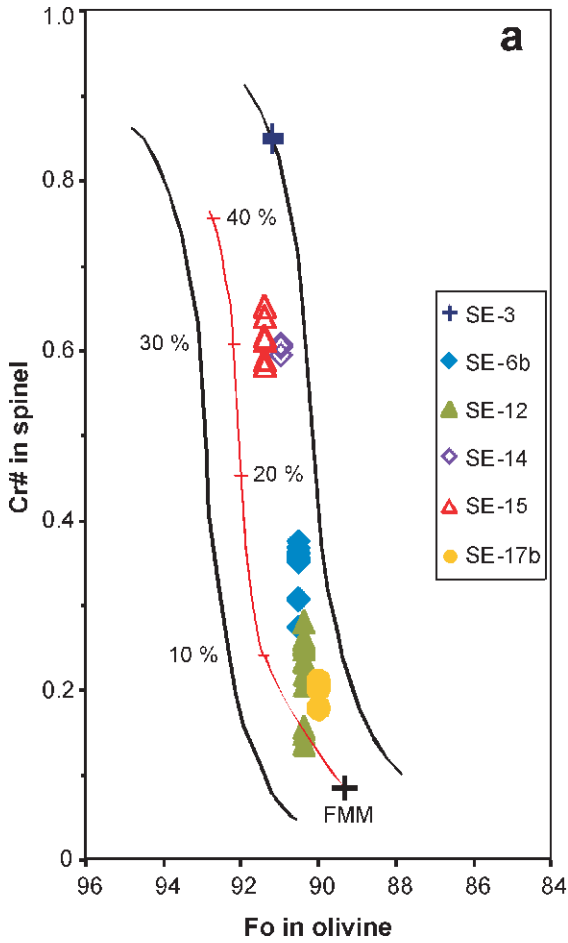
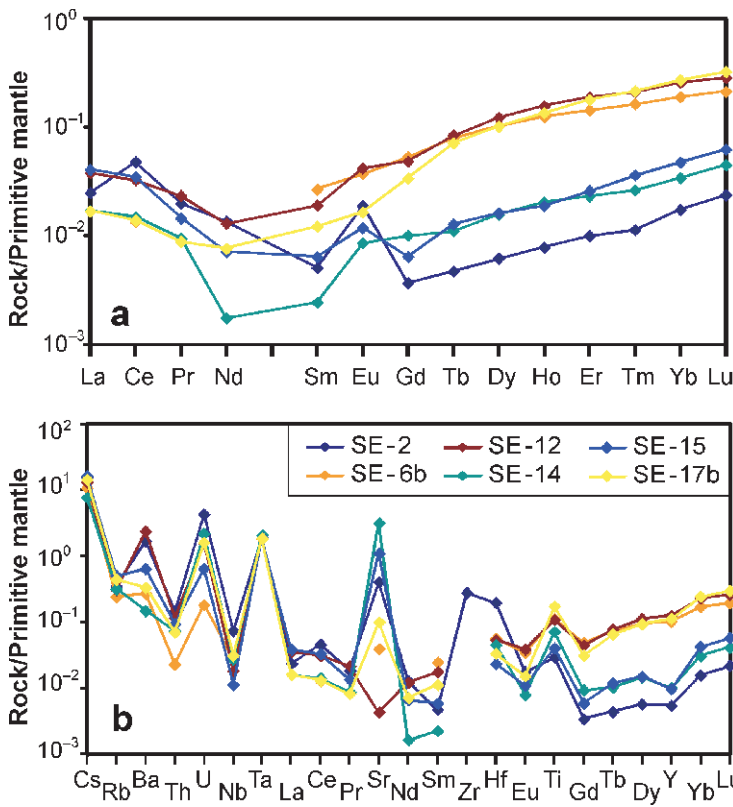


Fig. 8. a — Compositional relationship between Fo [=Mg/((Mg+Fe+Ca+Ni)/100) atomic ratio] content of olivine and Cr# [=Cr/(Cr+Al) atomic ratio] of spinel from the Sedlice peridotite body. OSMA (Olivine-Spinel Mantle Array) is a spinel peridotite residual trend and with melting trend (red curved line annotated by melting %) are from Arai (1987, 1994). FMM — fertile MORB mantle; **b** — Discrimination diagram Al₂O₃ versus Cr₂O₃ (wt. %) with plotted spinels from the Sedlice peridotite body (after Franz & Wirth 2000).



The PM normalized REE patterns indicate a depleted rock-suite, however an increase in LREE to HREE most likely reveals an influence of metamorphic fluids due to serpentinization. The whole-rock positive Eu anomaly might have been caused by substitution of Eu²⁺ for Ca²⁺ in tremolite or rare carbonates.

The low-temperature metamorphic mineral assemblage of peridotite body contains chrysotile, tremolite, rare andradite garnet, Cr-spinel; to chromite and magnetite, and rare carbonate; this could also be determined by an increase of the fayalite component in olivine. These features resemble the Meliatic meta-harzburgites in the Western Carpathians.

The Sedlice peridotite body most likely formed from the Meliatic Bôrka Nappe slices and currently appears to be a protrusive(?) body within the Central-Carpathian Paleogene sediments. The brittle fractures healed by magnesite and calcite crosscut the serpentinized peridotite body and

Fig. 9. a — Primitive Man; **b**— Spider diagram from Sedlice peridotite.

therefore may represent a younger carbonate mineralization of unclear age.

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